

# INVESTIGATION OF MECHANICAL PROPERTIES OF $Al_2O_3$ -20 WT % $ZrO_2$ COMPOSITES AS A FUNCTION OF SINTERING TEMPERATURE

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## Abstract

High purity nanocrystalline  $\alpha$ -alumina powder was mixed with 20 wt%  $ZrO_2$  by slurry method sintered at temperature ( $T_s$ ) 1450°C, 1500°C, 1550°C and 1600°C for 2 hour. The density, porosity, structural properties and mechanical properties of  $Al_2O_3$ - $ZrO_2$  composites with respect to sintering temperature have been explored in the present work. The XRD spectra indicate that  $\alpha$ - $Al_2O_3$ , t and m- $ZrO_2$  are the crystalline phases present in 20 wt% ZTA composites for all sintering temperature. It is observed that with higher sintering temperature the intensity of m- $ZrO_2$  phases increases and the t- $ZrO_2$  phases decreases. However, t- $ZrO_2$  retention becomes much easier to trigger the transformation to monoclinic. Higher density of 20 wt% ZTA has been achieved at 1600°C, whereas the highest porosity was obtained for sintering temperature 1450°C. The microstructures of the samples was studied by using SEM which represents highly homogeneous and finer structure at 1600°C. The effect of sintering temperature on hardness, elastic modulus and flexural strength were investigated and has been found to increase with the increase of sintering temperature. Maximum

hardness, elastic modulus and flexural strength were achieved for the Al<sub>2</sub>O<sub>3</sub>–20 wt% ZrO<sub>2</sub> composites sintered at 1600°C.

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**Keywords:** XRD; SEM; Vickers microhardness; Flexural strength.

## **Introduction**

A great progress in dental restorations technique has been established by the use of composite materials since the 70's. In some field application, the tendency is the elimination of the metallic substructure, including implant restorations. These composites show relative advantages, like esthetic, biocompatibility and chemical resistance. One problematic aspect of composite materials in particular is their low mechanical resistance and fracture toughness [Munro, 1997]. The metallic abutments used in prosthetic restorations with implants materials compromise the esthetic in some cases [Sadoun et. al., 1997]. To minimize this problem, some implant systems developed composite abutments. The use of alumina and zirconia with high density like biomaterial has been proposed. Alumina has shown excellent biocompatibility and wear resistance however exhibits low flexural strength and toughness [De Aza et. al., 2002]. The pure zirconia cannot be used in the manufacture of parts without the addition of stabilizers [Hannink, 2000]. The yttria-stabilized zirconia polycrystal has become a popular alternative to alumina as biomaterial and is used in dental applications such as endodontic posts [Fischer et. al., 2002], orthodontic brackets [Keith et. al., 1994], crowns and bridges [Cando, 2001 and Ardlin, 2002] and in composite abutments [Prestipino et. al., 1996], beside the extensive experience of using the composite in orthopedic implants [Christel et. al., 1989 and Serra et. al., 2002]. Zirconia presents good esthetic aspects after polishing, is inert in physiological environment, presents greater flexural resistance and toughness and lower Young's modulus when compared with pure alumina [Piconi et. al., 1999]. The ability of Zirconia Toughened Alumina (ZTA) to transform from tetragonal crystalline structure to a more voluminous monoclinic structure and thus obstruct crack propagation, gives the material its strength and toughness [Ardlin, 2002]. The addition of zirconia in the alumina have being used for long time with the objective of alumina based composites densification. It is well known that the mechanical properties of alumina composites can be considerably increased by the incorporation of fine zirconia particles [Claussen, 1976, Shin et. al., 1990, Mullot et. al., 1993, Kishino et. al., 1994]. The toughening mechanisms associated with zirconia toughened alumina are mainly based on the stress-induced tetragonal→monoclinic martensitic transformation toughening.

The main objective of the present work is to incorporate 20 wt% ZrO<sub>2</sub> particles synthesized by the slurry method into high purity alumina

matrix and to investigate the effect of ZrO<sub>2</sub> addition on the microstructures and mechanical properties of the Al<sub>2</sub>O<sub>3</sub> composites at different sintering temperature.

## **Experimental**

### **Sample Preparation**

The commercially available powders used in this work were the nanocrystalline,  $\alpha$ - Al<sub>2</sub>O<sub>3</sub> powder and yttria-stabilized ZrO<sub>2</sub> (3 mol%) powder from the manufacturing company Inframat<sup>®</sup> Advanced Materials<sup>™</sup>, USA. Alumina presented crystal size ~40 nm and particle size ~150 nm and zirconia presented average particle size 30-60 nm. To prepare composition, related powders were weighed by following unitary method, keeping the total weight constant.

The nanocrystalline,  $\alpha$ - Al<sub>2</sub>O<sub>3</sub> powder and 20 wt % yttria-stabilized ZrO<sub>2</sub> (3 mol %) powder have been taken for the preparation of required composite. To attain the composition, mechanical milling was performed in distilled water with 65 wt% of solid content for 24 hours in ball milling, with zirconia ball as a grinding media. Ammonium polyacrylate was used as dispersant. The slurry of the powder mixtures was dried in an oven for 24 hours at 100°C and the dried lumps were crushed and passed through a plastic sieve. Hand milling was done for several hours to obtain a homogeneous distribution. Powder compacts with dimension of 5×6×50 mm were confirmed by uniaxially pressing at 60 MPa. Before compaction, polyvinyl alcohol (PVA) was mixed as a binder with the powder to provide some green strength for subsequent handling. The withdrawal of the binder was carried through in a muffle furnace with heating rate 6 °C/min at 600 °C per 180 min. The sintering was carried out in a box furnace at 1450°C, 1500°C, 1550°C and 1600°C for two hours with a heating rate of 20°C/min. The bulk density of the sintered sample was measured from the ratio weight/volume. The sintered specimen was machined longitudinally with a 325 grit-resin bonded diamond wheel at a depth of 5 µm/pass. The final dimension of the specimens was 3×4×35 mm.

### **Measurements**

Phase identification was confirmed by powder X-ray diffractometry (XRD) technique using BRUKER X-Ray Diffractometer of Model D8 Advance with Cu-K<sub>α</sub> radiation ( $\lambda = 1.54 \text{ \AA}$ ). For each sample a scan has been performed from 20° to 75° with a step size of 0.02°. The volume fraction of monoclinic zirconia ( $V_m$ ) was calculated according to the following equation:

$$V_m = \frac{I_{(\bar{1}11)_m} + I_{(111)_m}}{I_{(\bar{1}11)_m} + I_{(111)_m} + I_{(111)_i}}$$

Theoretical density  $\rho_o$  of the sample was calculated from the actual density of the starting powders and their weight percentages using the rule of mixtures [Bansal et. al., 2003]:

$$\rho_o = \rho_A V_A + \rho_Z V_Z$$

where  $\rho_A$ ,  $\rho_Z$  are the density and  $V_A$ ,  $V_Z$  are the volume fraction of  $Al_2O_3$  and  $ZrO_2$  respectively. Values of  $\rho_A = 3.97 \text{ gm/cm}^3$  and  $\rho_Z = 6.10 \text{ gm/cm}^3$  have been used here to calculate  $\rho_o$ . The measured bulk density  $\rho_b$  was determined using the formula:

$$\rho_b = \frac{m}{\pi r^2 h}$$

Where  $m$  is the mass,  $r$  the radius and  $h$  the height of the sample.

The porosity  $P$  of the sample was determined using the relation.

$$P = \left(1 - \frac{\rho_b}{\rho_o}\right) \times 100\%$$

Microstructures for the samples of different compositions were examined by using Philips XL-30 scanning electron microscopy (SEM). Mean linear intercept length  $\bar{l}$  is used to determine the ASTM grain size number,  $G$ , by using the linear intercept method:

$$G = -3.2877 + 6.6439 \log_{10} \bar{l}$$

where  $\bar{l}$  is in millimeters at  $1 \times$  Magnification.

The methodology used for the determination of hardness, was in accordance with ASTM C1327-99 [Cecilia et. al., 2004]. Five Vickers impressions had been carried through in the surfaces of each one of the samples, which already were polishing, using an applied load of 2 kgf (19.6141 N) during six seconds. After the diagonal length measurement, the values of the Vickers hardness (GPa) were calculated, by the following equation [Cecilia et. al., 2004, Santos et. al., 2007]:

$$H_v = 0.018544 \times \frac{F}{d^2}$$

Where  $F$  is the load in N and  $d$  is the average length of the two diagonals of the indentations in mm.

The porosity dependence Elastic Modulus can also be measured using the Dewey-Mackenzie relation [Pabst et. al., 2004]:

$$E = E_o (1 - 2P)$$

where  $E$  is the effective elastic modulus of the porous composites,  $E_o$  is the elastic modulus of the dense composites and  $P$  is the porosity of the sample.

**The flexural strength of the specimens was determined according with ASTM D790–10 by three-point bending method (Hounsfield Model H10KS-universal test machine), with a span of 30 mm and crosshead speed of 0.5 mm/min at ambient room temperature conditions. The flexural strength was calculated from the following equation.**

$$\sigma_{flexural} = \frac{3WL}{2BD^2}$$

where W is the load (N), B is the width (mm) and D is the thickness (mm) of the sample.

## Result and Discussion

### XRD analysis

The XRD patterns of a ZTA sample containing 20 wt% ZrO<sub>2</sub> sintered at 1450°C, 1500°C, 1550°C and 1600°C are shown in Figure 1, which confirms the formation of different phase in ZTA.

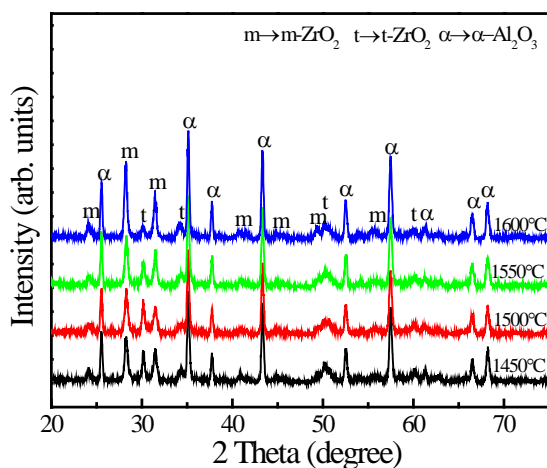


Figure 1. XRD pattern of Al<sub>2</sub>O<sub>3</sub>-20 wt% ZrO<sub>2</sub> composite sintered at different temperature

XRD analysis of the samples indicates that  $\alpha$ -Al<sub>2</sub>O<sub>3</sub>, t and m-ZrO<sub>2</sub> are the phases present in the composite. From the figure it is observed that with the increase of sintering temperature the intensity of the peaks for Al<sub>2</sub>O<sub>3</sub> and ZrO<sub>2</sub> have been changed. It also reveals that with the increase of sintering temperature, the intensity of m-ZrO<sub>2</sub> phases increases, whereas the intensity of t-ZrO<sub>2</sub> decreases. However, t-ZrO<sub>2</sub> retention becomes much easier to trigger the transformation to monoclinic. As a result, its contribution to transformation toughening predominates. The extent of toughening achieved in the composites depend on the grain size of Al<sub>2</sub>O<sub>3</sub> and ZrO<sub>2</sub>, volume fraction of ZrO<sub>2</sub> retained in the metastable tetragonal phase as well as on the relative distribution of Al<sub>2</sub>O<sub>3</sub> and ZrO<sub>2</sub> in the matrix [Bleier et. al., 1992].

Finer particle size of both  $\text{Al}_2\text{O}_3$  and  $\text{ZrO}_2$  will not only enhance the chances of a uniform  $\text{Al}_2\text{O}_3$  and  $\text{ZrO}_2$  distribution, it also increases the possibility of  $\text{ZrO}_2$  being retained as metastable tetragonal phase [Garvie, 1965]. So it can be inferred that monoclinic phase is indeed retained at the room temperature due to the presence of the hard alumina matrix around and this phase is responsible for flexural strength improvement by transformation toughening mechanism.

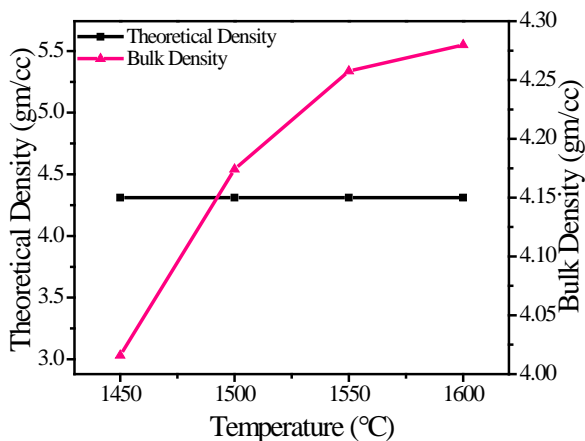


Figure 2. Effects of the sintering temperature on the theoretical and bulk density of  $\text{Al}_2\text{O}_3$ -20 wt%  $\text{ZrO}_2$  composite

Figure 2 shows the result of density as a function of sintering temperature. With the continual supply of thermal energy, more and more contact points are created between the particles and the intervening porosity diffuses out of the contact area. Hence a substantial increase in density is resulted. At 1600°C, higher density is achieved than that of other temperature. Densification of powdered particle is a thermally assisted phenomenon due to sintering at various temperatures. We know that matter transport during the sintering of polycrystalline composites occurs by diffusion, a thermally activated process, which can occur along different paths in the solid, giving rise to the different mechanisms of diffusion: lattice, grain boundary and surface diffusion. The density is also influenced by phase transformation of monoclinic to tetragonal zirconia above 1170°C during heating. The density result reveals that the elimination of pore, simultaneous grain growth with grain boundary diffusion and lattice diffusion are the predominant mechanism to compact the material at high temperature. In this work, the relative density have been calculated about 93.15% for 1450°C and about 99.28% for 1600°C. The relative density for other temperatures are in between this values.

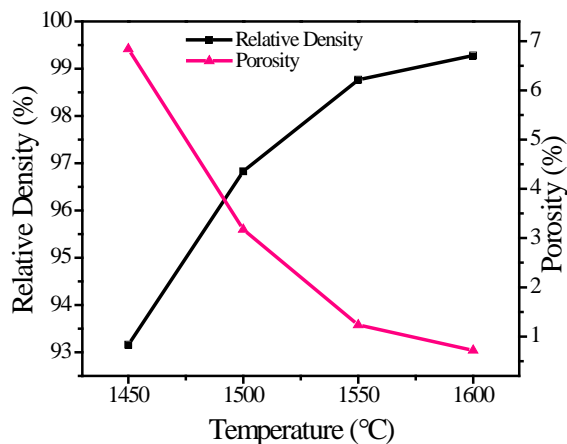


Figure 3. Effects of the sintering temperature on the relative density and porosity of Al<sub>2</sub>O<sub>3</sub>-20 wt% ZrO<sub>2</sub> composite

About 1-4% density has been changed due to increase of 50°C temperature. This resembles that after increasing sintering temperature particles become more compact and dense, therefore for the same reason porosity decreases with the same change of temperature. Moreover, increasing temperature, density increases and porosity decreases regularly than that of lower temperature that has been shown in Figure 3. It is observed that the porosity is decreased with sintering temperature. It resembles that the composite are in more compact form at the higher temperatures. During the diffusion process of sintering, the intervening porosity diffuses out, resulting in reduction of porosity. It is observed that the porosity is decreased with sintering temperature. This porosity usually originates from entrapped pore during green slurry. With the increase of temperature, sample becomes more homogeneous. During 50°C interval, porosity decreases sharply from low temperature to high temperature. It is also seen that the percentage of porosity is minimum at the sintering temperature 1600°C for the composites than that of at 1450°C, i.e., it reveals that the sample is more compact at this temperature. The calculated density values are in good agreement with some previous results obtained using pressureless sintering technique [Maca et. al., 2008, Bodisova et. al., 2007, Oungkulsolmongkol et. al., 20010]. The optimum density was also achieved for ZTA sintered at 1600°C for the composition.

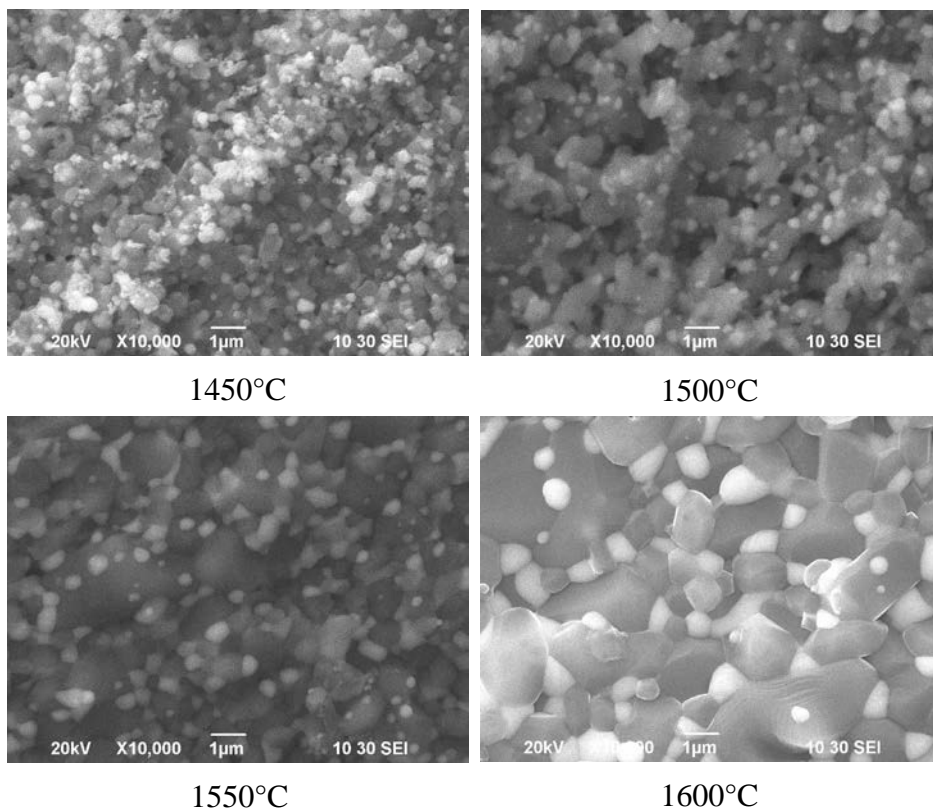


Figure 5. Microstructure of  $\text{Al}_2\text{O}_3$ -20 wt%  $\text{ZrO}_2$  composite sintered at different temperature

### ***Scanning electron microscopy***

Figure 5 show the SEM micrographs of the ZTA composites containing 20 wt%  $\text{ZrO}_2$  at different temperatures. Micrographs reveals highly homogeneous microstructures with some agglomerates, pores or abnormally grown alumina grains. The micrograph exhibits the  $\text{Al}_2\text{O}_3$  and  $\text{ZrO}_2$  grains as dark and whitish color respectively, where zirconia grains are embedded around the alumina grain uniformly and within the grain. However, small fractions of intragranular zirconia grain are also being noticed.

From the figure, it can also be observed that  $\text{ZrO}_2$  particles are uniformly dispersed throughout the alumina matrix. As the temperature increases, the  $\text{Al}_2\text{O}_3$  and  $\text{ZrO}_2$  grain size increases significantly. The  $\text{ZrO}_2$  phase creates a pinning effect around  $\text{Al}_2\text{O}_3$  grain and obstructs its growth. It can also be observed that the grain sizes of  $\text{ZrO}_2$  increase and hinders the grain growth of  $\text{Al}_2\text{O}_3$  at different temperatures, which contributes to the lowering of porosity. A fine grain size and refined microstructure are



necessary for improved mechanical properties especially for wear resistance of dental implant. It is observed from Figure 6 that, grain size of  $\text{Al}_2\text{O}_3$  and  $\text{ZrO}_2$  vary with the sintering temperature. In the present work, it is seen that as the sintering temperature was increasing, then the grain size increases regurly.

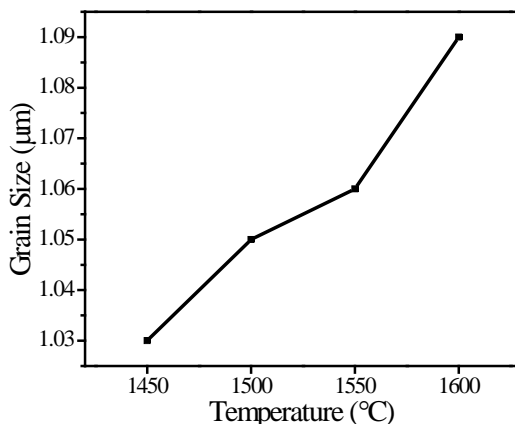


Figure 6. Effects of the sintering temperature on the grain of  $\text{Al}_2\text{O}_3$ -20 wt%  $\text{ZrO}_2$  composite

### ***Hardness and elastic modulus***

Figure 7 shows the variation in Vickers hardness of 20 wt% ZTA composite with the change of temperature. Hardness increases with the increase in temperature. As higher sintering temperature offers better densification, hardness value increases as a whole. Similar trend in hardness has been observed by R. P. Rana [Rana, 2009]. Similar trend in hardness has also been observed by Basu et al. [Basu et. al., 2005]. At 1600°C, maximum hardness shows 11.52 GPa. However, at 1450°C, minimum hardness shows 9.61 GPa. The above increase in the hardness value of the  $\text{Al}_2\text{O}_3$ - $\text{ZrO}_2$  composite is related to the density increment. It appears that the phases of the sintered composite have least influence on the hardness.

Elastic modulus of the composite depends on the volume fraction of the phases present and their individual modulus of elasticity. At 1450°C, densification of the composites is improved by reducing pore volume fraction of the matrix and as a result of which the modulus of elasticity improves. Alumina has a higher modulus of elasticity compared to zirconia. So, samples containing 20 wt% zirconia have the highest modulus of elasticity at 1600°C. Simultaneously, the grain size of zirconia increases and hinders the grain growth of alumina at 1600°C, which contributes to the change of elastic modulus. Nicholson [Khaund et. al., 1980] established that both Young's modulus and hardness of ZTA composites containing particles of zirconia stabilized with 0.3 mol% MgO followed a linear rule of mixtures.

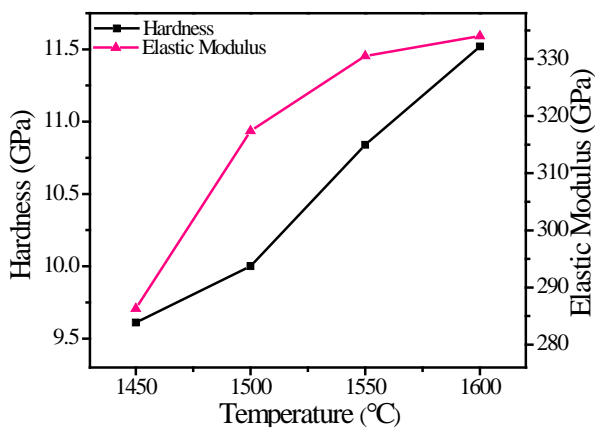


Figure 7. Effects of the sintering temperature on the relative density and porosity of  $\text{Al}_2\text{O}_3$ -20 wt%  $\text{ZrO}_2$  composite

### ***Flexural Strength***

The Flexural strength increases almost linearly with increasing sintering temperature that has shown in Figure 8. The flexural strength was found around 728 MPa sintered at 1600°C and that of 688 MPa sintered at 1450°C for 2 hours.

The martensitic transformation and frontal process zone are predominant mechanisms to increase the flexural strength of  $\text{Al}_2\text{O}_3$ - $\text{ZrO}_2$  nanocomposite. Since the tetragonal zirconia grains do not undergo the stress-induced transformation to the monoclinic form when grain size is smaller than a critical size. The particle sizes lower than the critical value does not affect the strength. The reduction of size improves the bulk density but not the transformability of the tetragonal zirconia grains. Usually, the effect of grain size on strength in transformation-toughened composites is a complex phenomenon [Orange et. al., 1988]. An increase in grain size improves the transformability, therefore, increasing strength. Microstructure reveals that the zirconia grain size increases with increasing zirconia content. The m- $\text{ZrO}_2$  phase increases with increase in sintering temperature. The retention of t- $\text{ZrO}_2$  varies during fracture and t- $\text{ZrO}_2$  fraction decreases with increasing temperature. However, t- $\text{ZrO}_2$  retention becomes much easier to trigger the transformation to monoclinic. As a result, its contribution to transformation toughening predominates although retention of m- $\text{ZrO}_2$  aids to microcrack toughening.

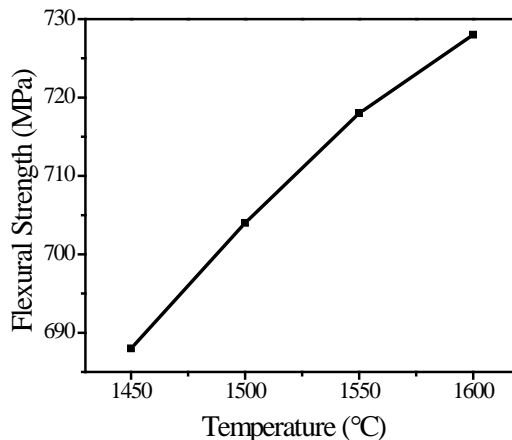


Figure 8. Effects of the sintering temperature on the flexural strength of  $\text{Al}_2\text{O}_3$ -20 wt%  $\text{ZrO}_2$  composite

## Conclusion

This work presents the variation in structural and mechanical properties of 20 wt% ZTA in the temperature range from 1450°C to 1600°C. All samples in this work have been prepared by slurry method. The X-ray diffraction studies clearly showed the formation of  $\alpha$ - $\text{Al}_2\text{O}_3$ , t and m- $\text{ZrO}_2$  phases in the ZTA composites. The density has been found to increase with increasing sintering temperature, while porosity decreases with the same change of temperature. SEM images show the microstructures of prepared ZTA which reveals that the grains are homogeneous and uniformly distributed. It can also be observed that the grain sizes of  $\text{ZrO}_2$  increase and hinders the grain growth of  $\text{Al}_2\text{O}_3$  at 1600°C compared to that of at 1450°C, which contributes to the lowering of porosity. Hardness and elastic modulus was found to be increased with sintering temperature. The Flexural strength increases with sintering temperature. The predominant mechanisms for the improvement of the mechanical properties of the ZTA materials are the stress-induced transformation toughening and process flaws eliminating. The change of sintering temperature was also found to have a significant effect on structural and mechanical properties of ZTA composite.

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